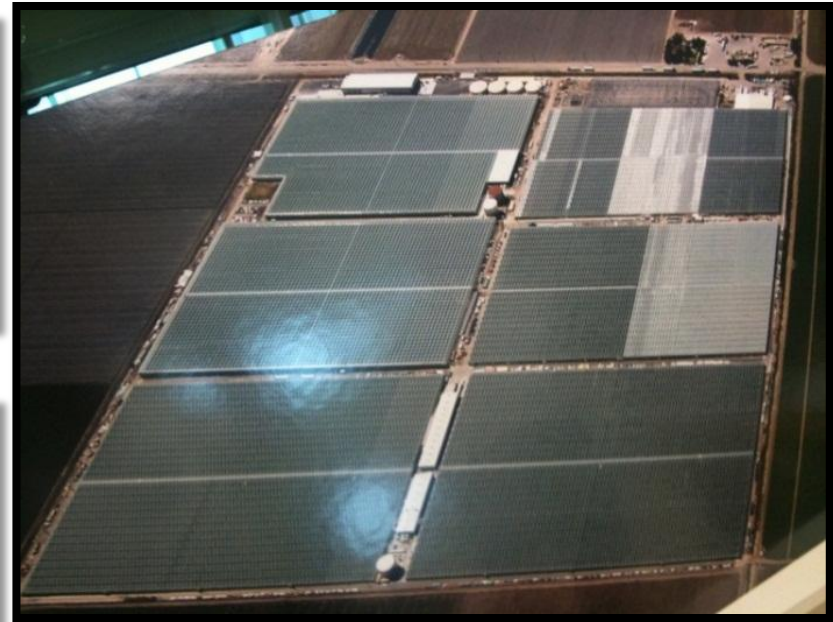


Produce Water Treatment



Horticulture Water Treatment



Water Education Alliance
For Horticulture

WaterEducationAlliance.org

Typical Pathogens

- ▣ Generic *E. Coli*
- ▣ *E. Coli* O157:H7
- ▣ *Salmonella*
- ▣ *Listeria*
- ▣ *Legionella*
- ▣ *Pseudomonas*
- ▣ *Cryptosporidium*
- ▣ *Norwalk Virus*
- ▣ *Giardia Lamblia*
- ▣ *Cyclospora*

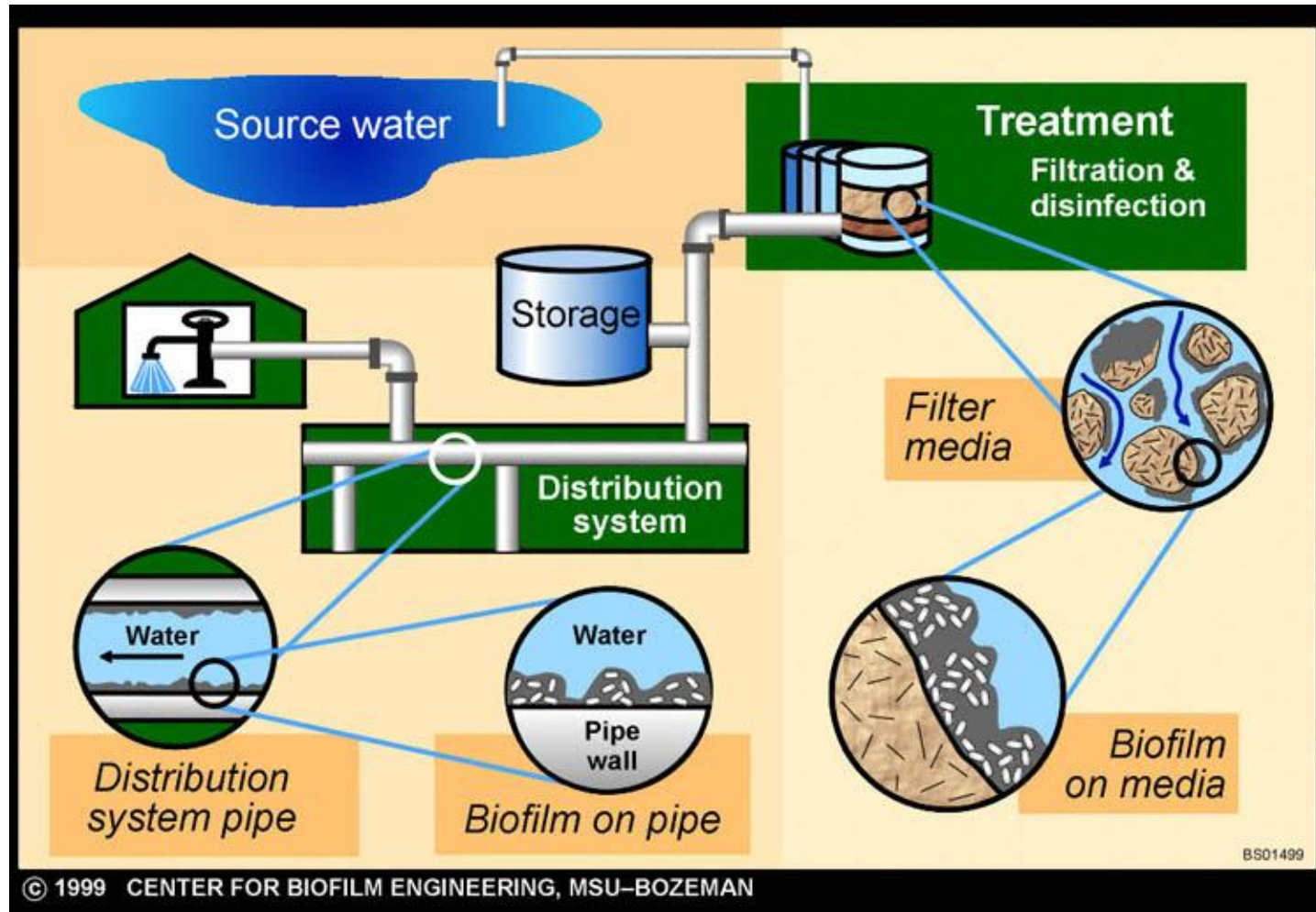
Pathogen Issue

- ❑ Free floating in suspension
- ❑ Trapped in Biofilm
- ❑ Biofilms grow in water distribution systems
- ❑ Biofilms are a harborage and growth medium
- ❑ Biofilms must be Killed *and* Removed

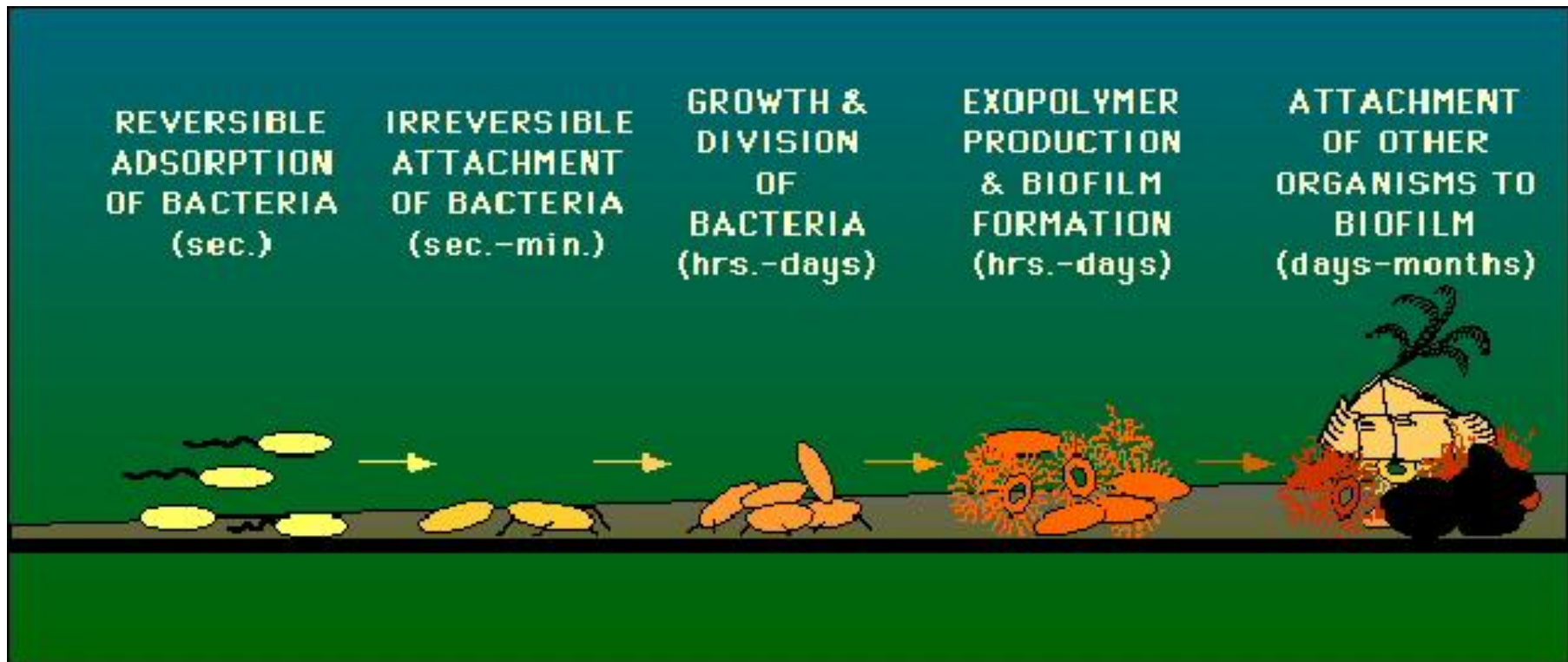
Direct Benefits

- ❑ Remove and Kill Biofilm
- ❑ Microbial Reduction
- ❑ Prevent Product Contamination
- ❑ Improve Quality
- ❑ Reduced Chemical Use
- ❑ Consistent Water Flow
- ❑ Reduced equipment and pipe corrosion
- ❑ Reduced mineral contamination
- ❑ Reduce lost time and labor for sanitation
- ❑ Reduce deposits of scale and biofilm
- ❑ Loss of heat transfer and higher energy cost

Biofilm Development

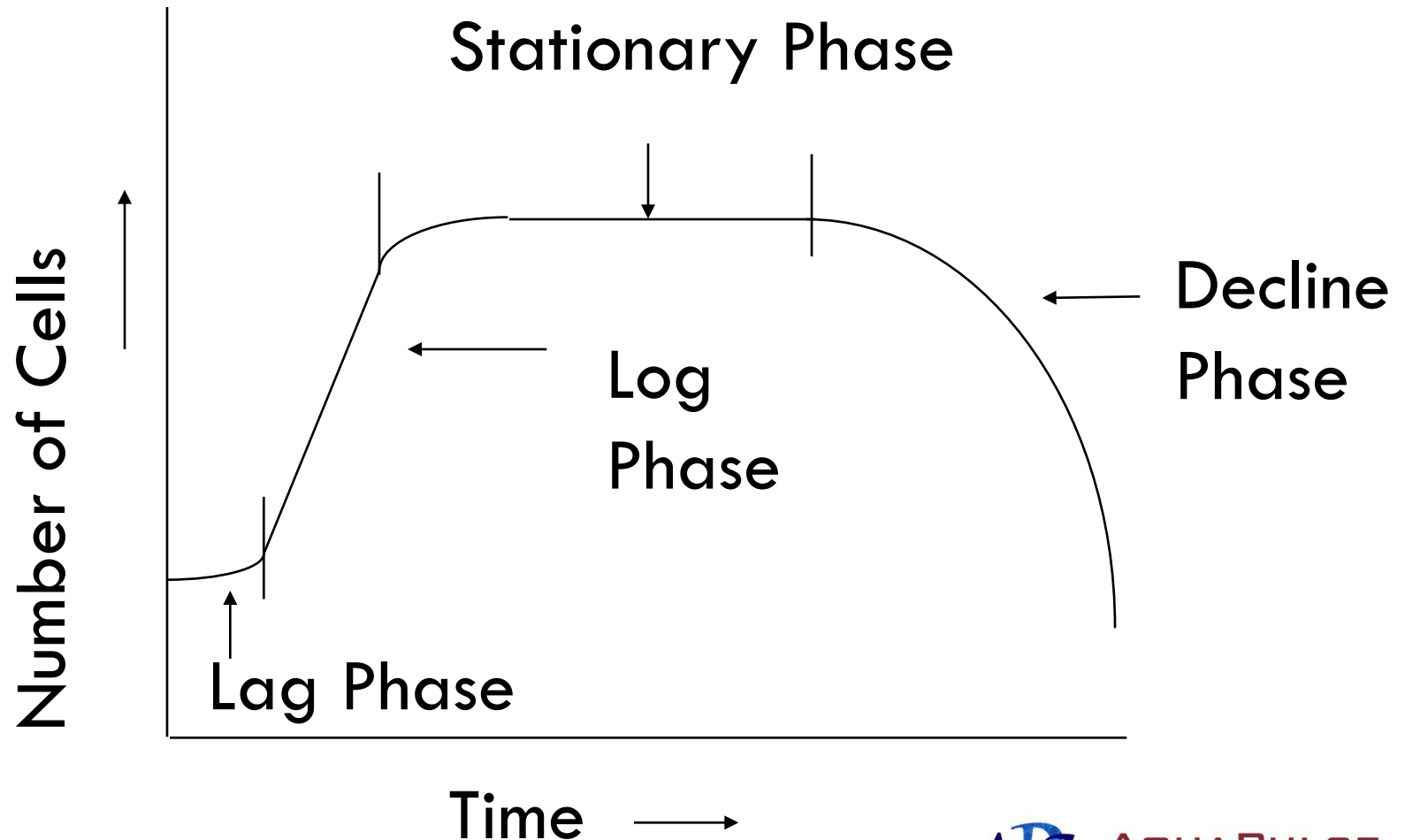


Biofilms

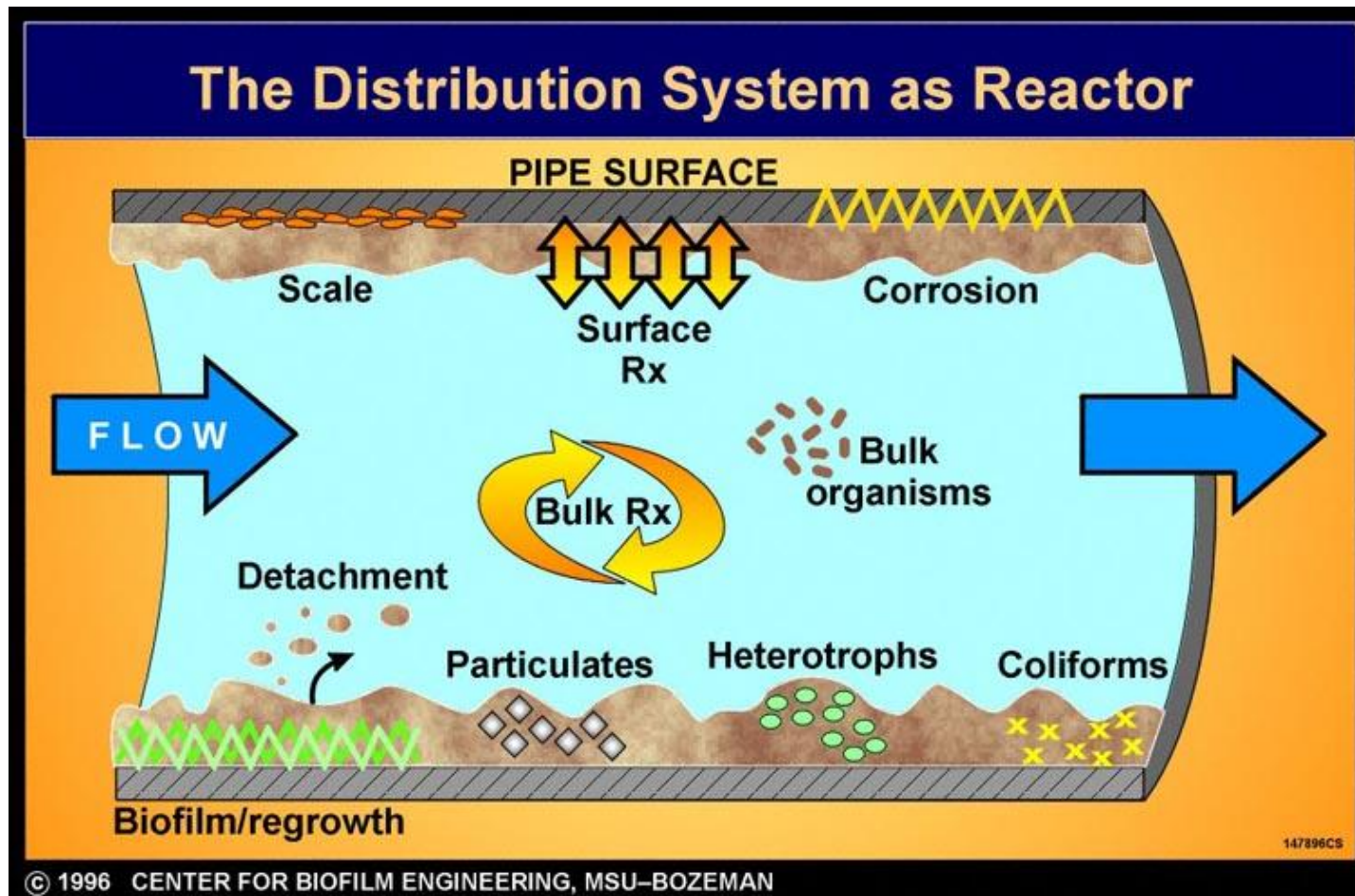


John Lennox, 2008 (<http://www.personal.psu.edu/faculty/j/e/jel5/biofilms/>)

Growth Curve

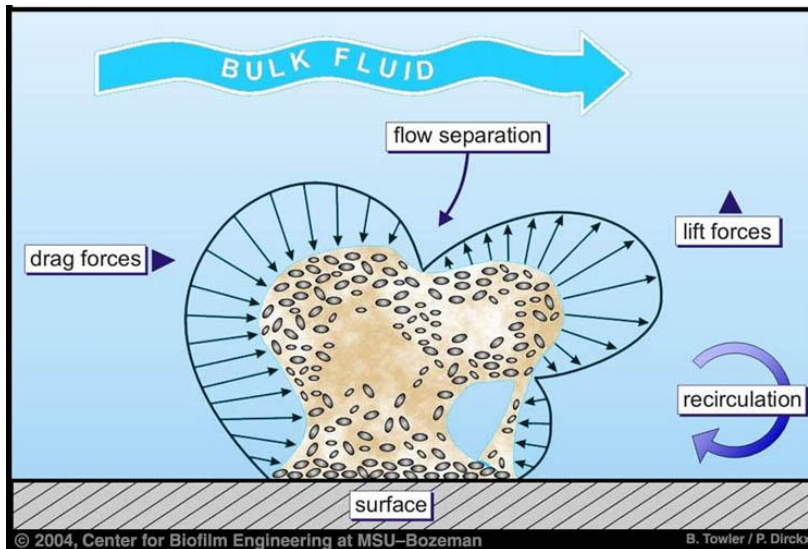


Biofilm in Distribution Pipes

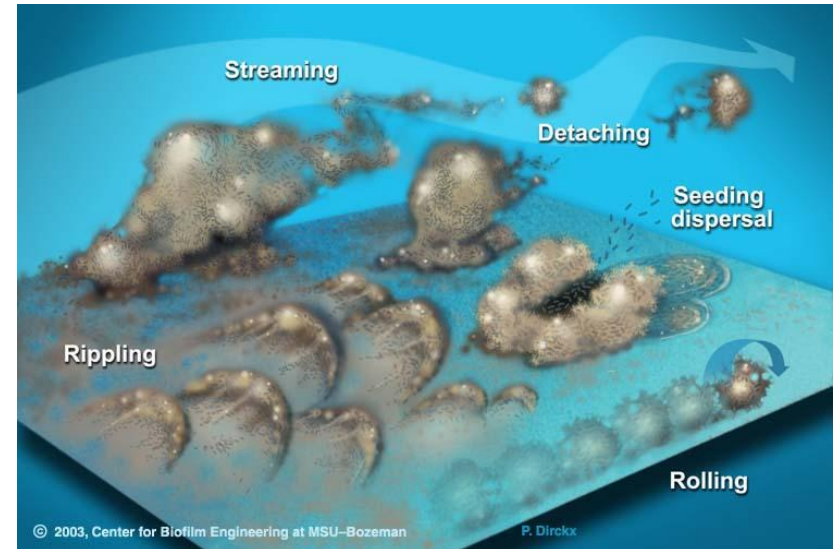


Biofilm in Water Flow

Fluid Dynamics

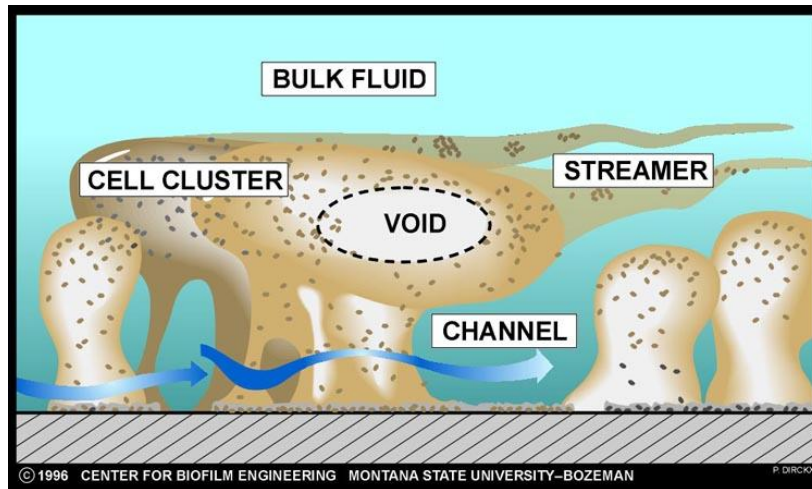


Biofilm in Water Flow

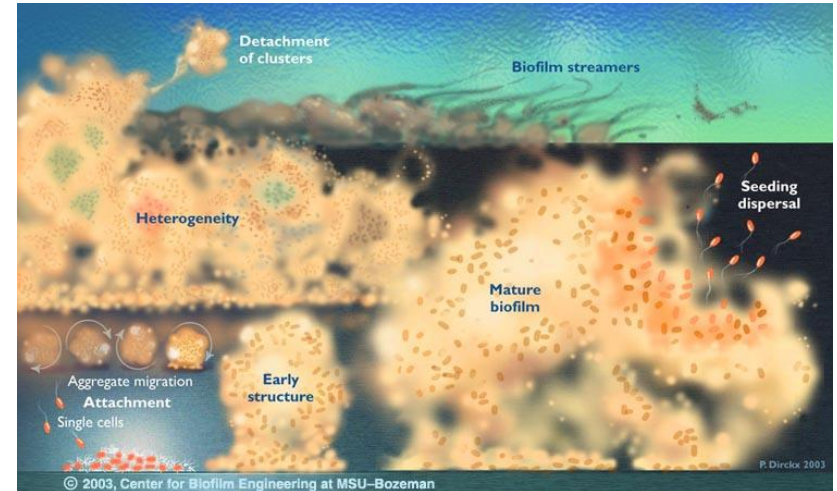


Biofilm Characterization

Biofilm Dynamics



Biofilm Environment



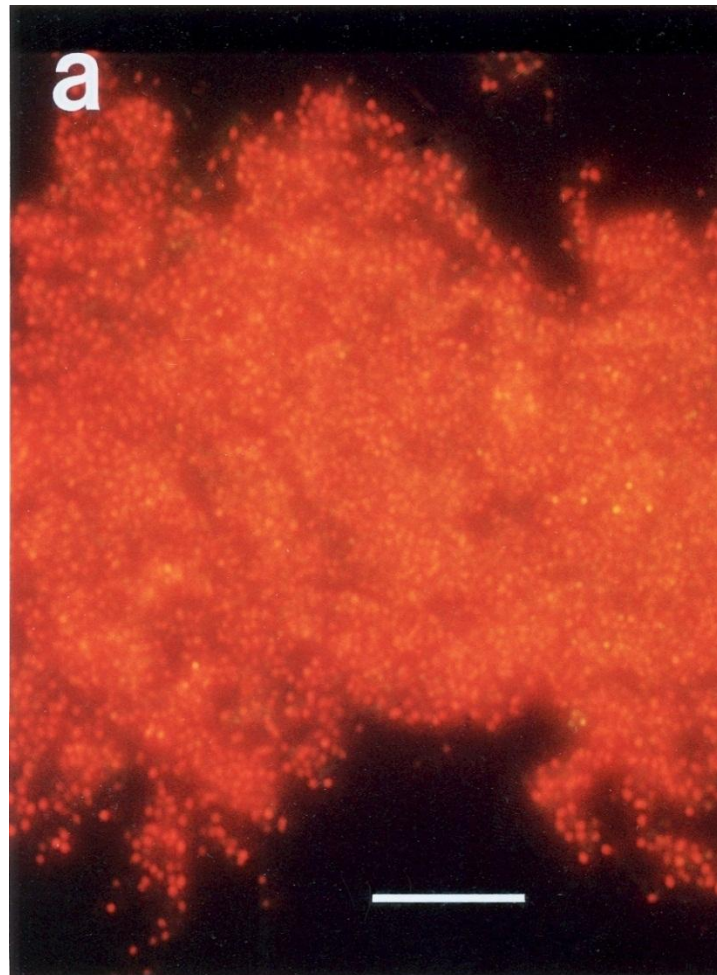
Pathogens *grow* in Biofilms

- Listeria and Facultative anaerobes
 - ▣ Grows in the presence & absence of oxygen
- Temperature
 - ▣ Growth 32 – 113°F
 - Growth at pH 4.4 - 9.6
 - Optimum pH 6.5-7.5
 - Survival can occur at pH <4.3
 - pH tolerance is temperature and acid dependent

Respiratory activity inside biofilm

Respiratory
activity

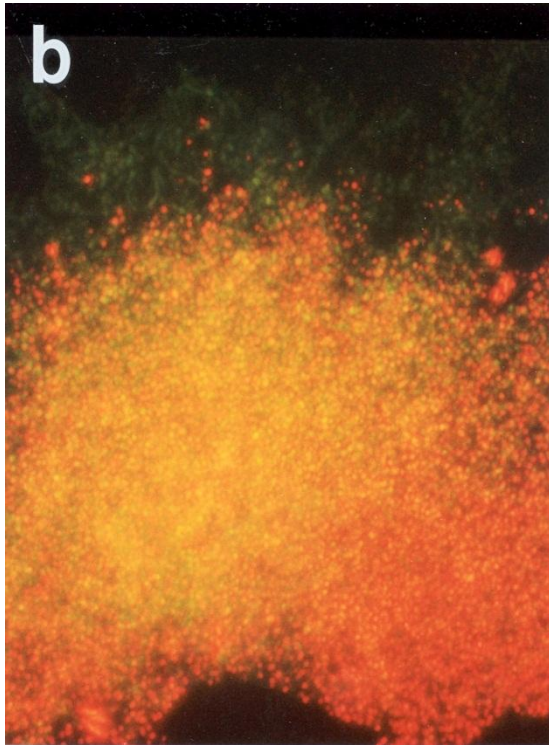
Untreated Control



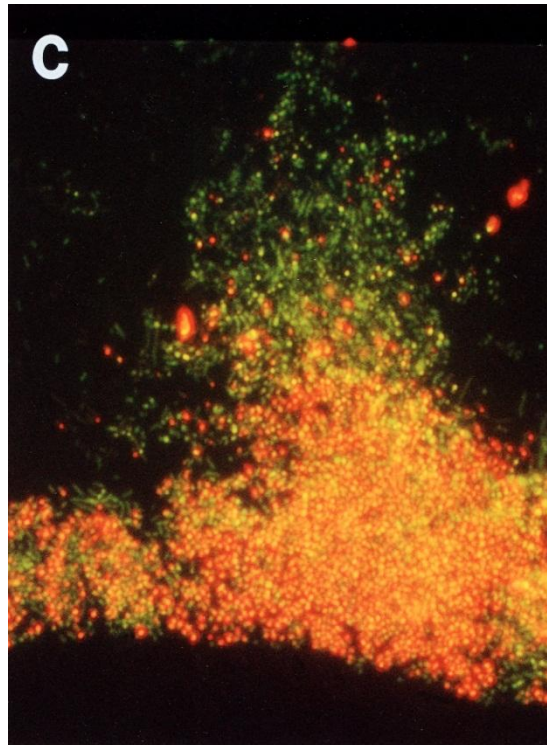
Huang et al (1995)
*Applied
Environmental
Microbiology*
61:2252.

Respiratory activity inside the Biofilm

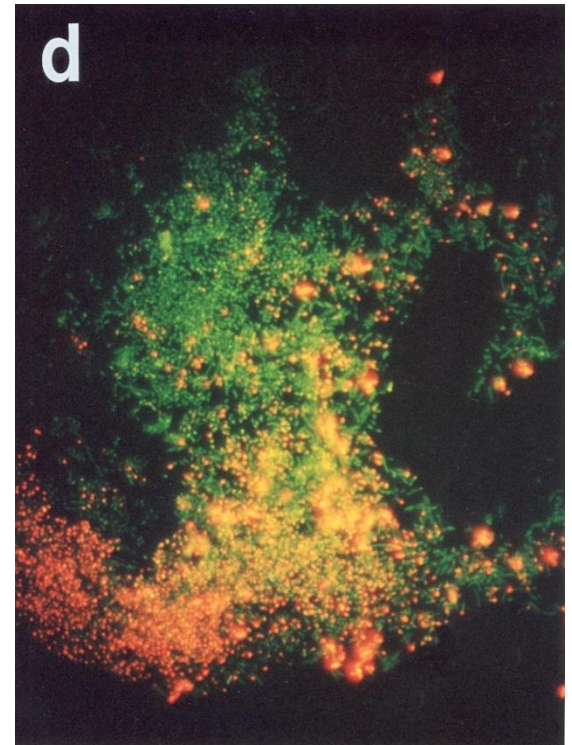
Chemical Treatment



After 30 min



After 60 min

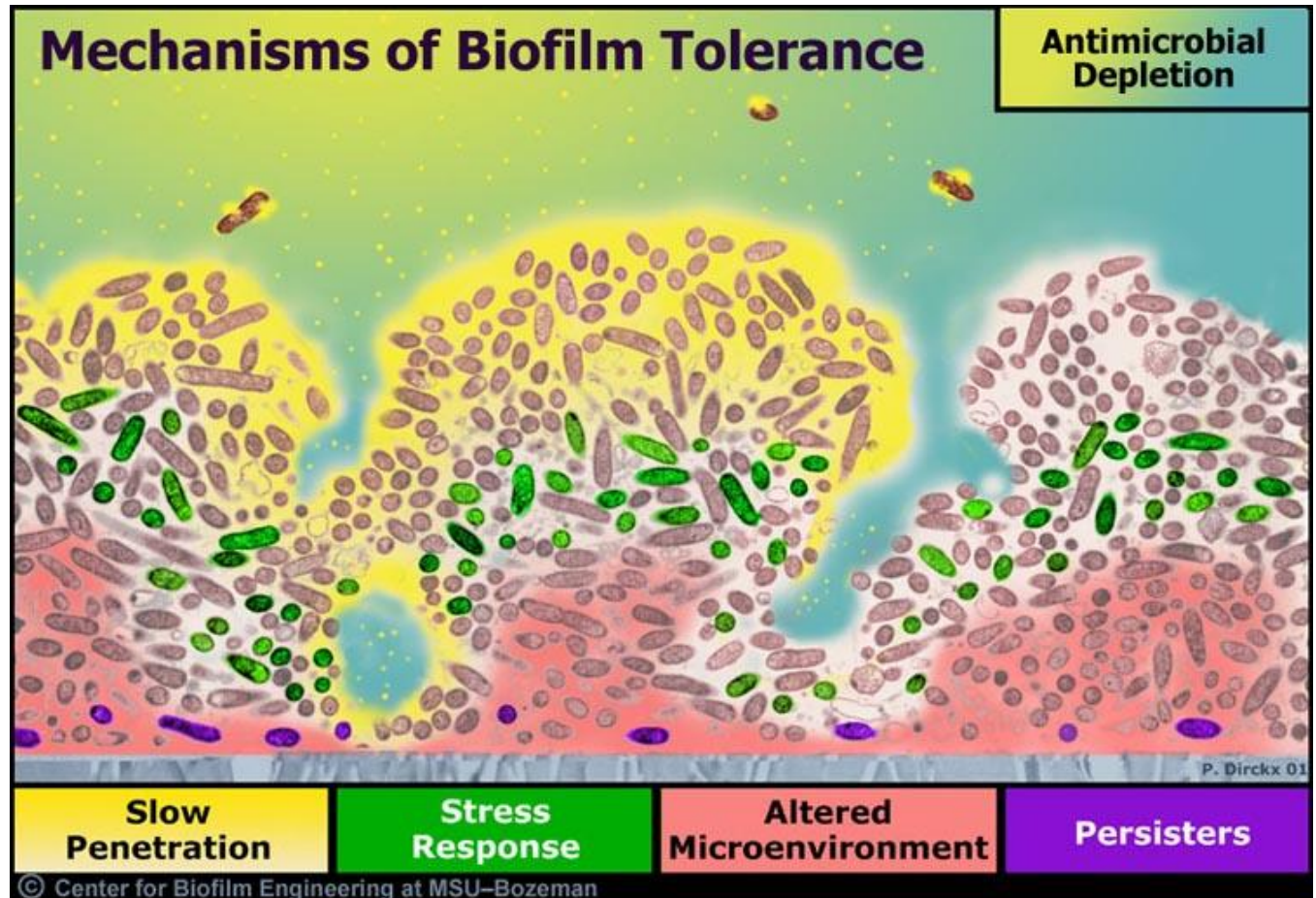


After 120 min

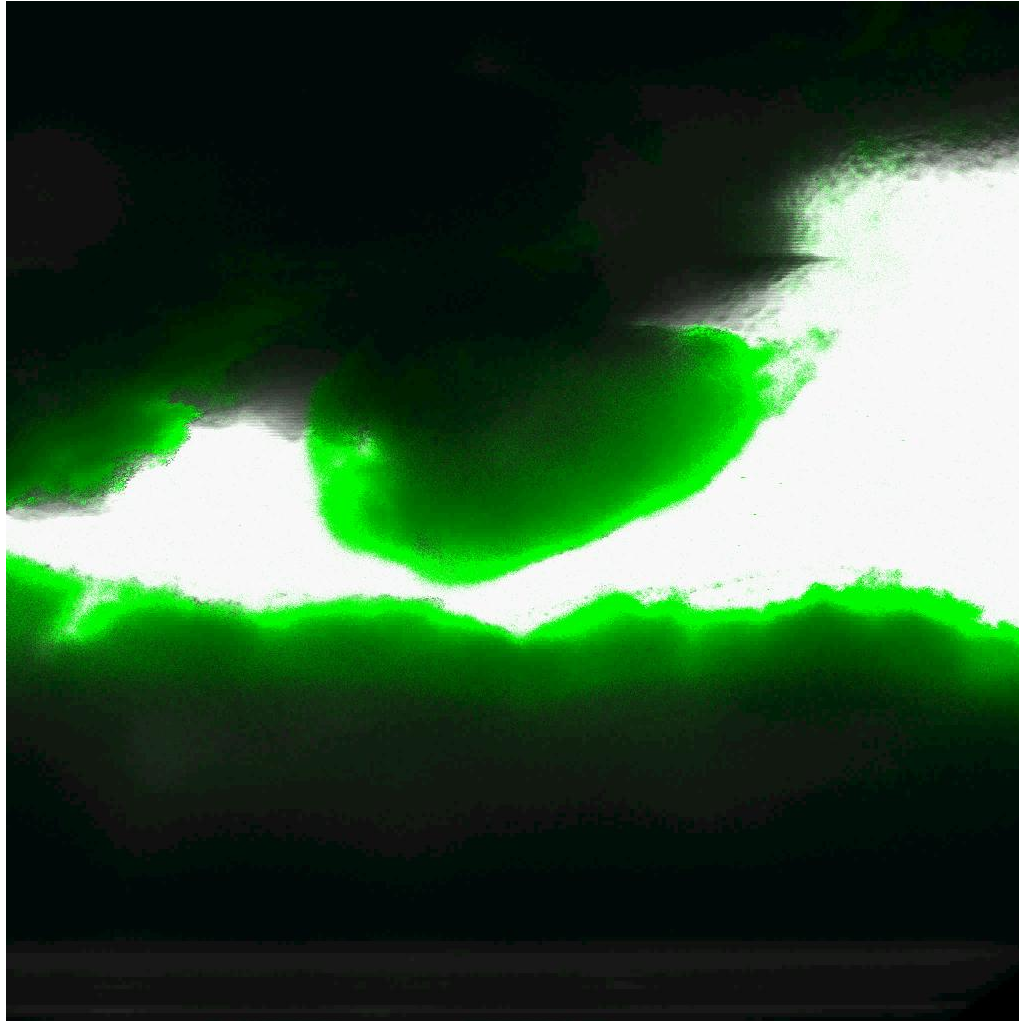
Huang et al (1995) *Applied Environmental Microbiology* **61**:2252.

Biofilm Sanitizer Treatment

Treatment of Biofilms with chlorine water sanitizers result in poor penetration

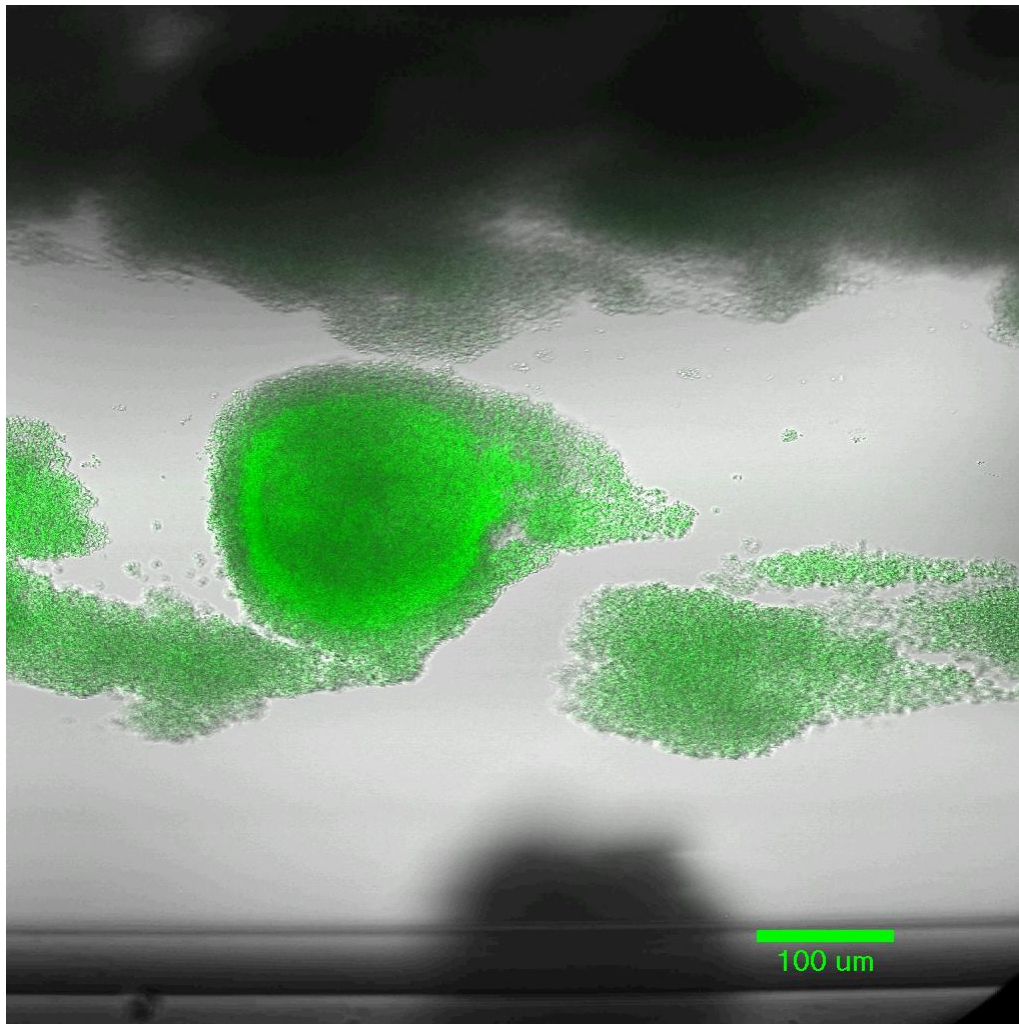


Treatment with 50 mg/l Quat



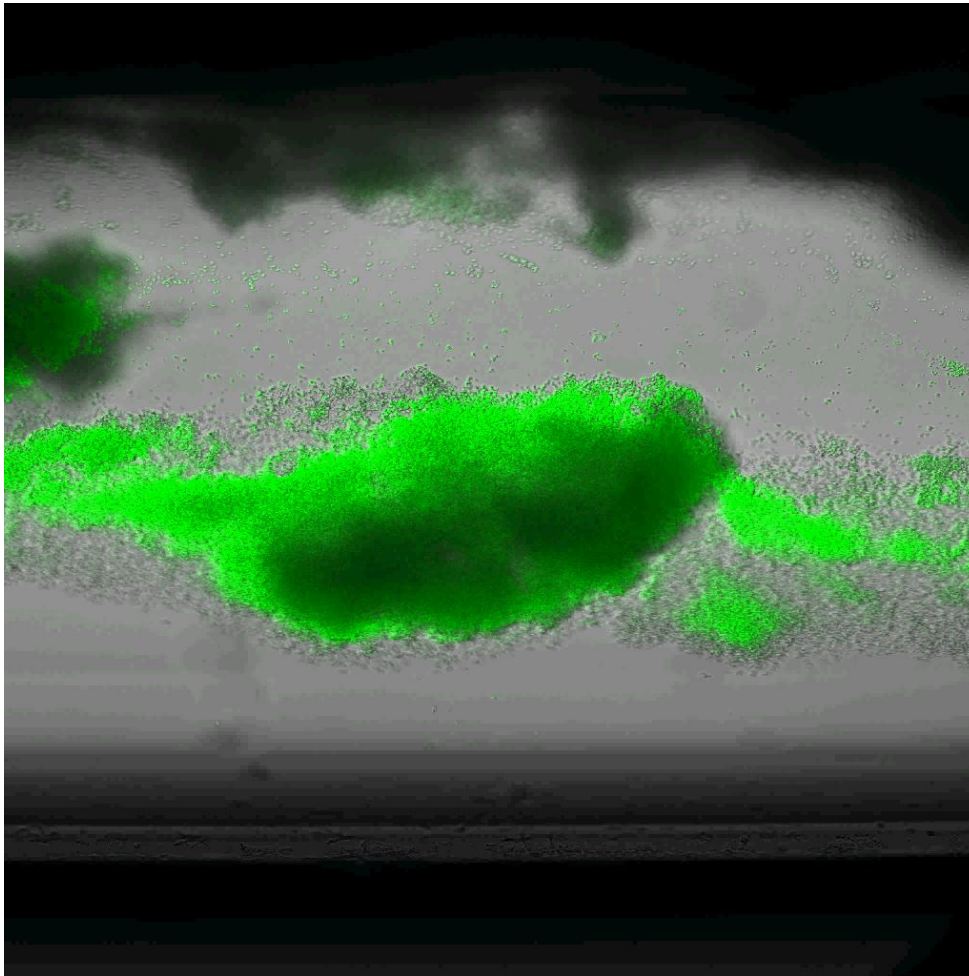
60 min

Treatment with 50 mg/l Nisin



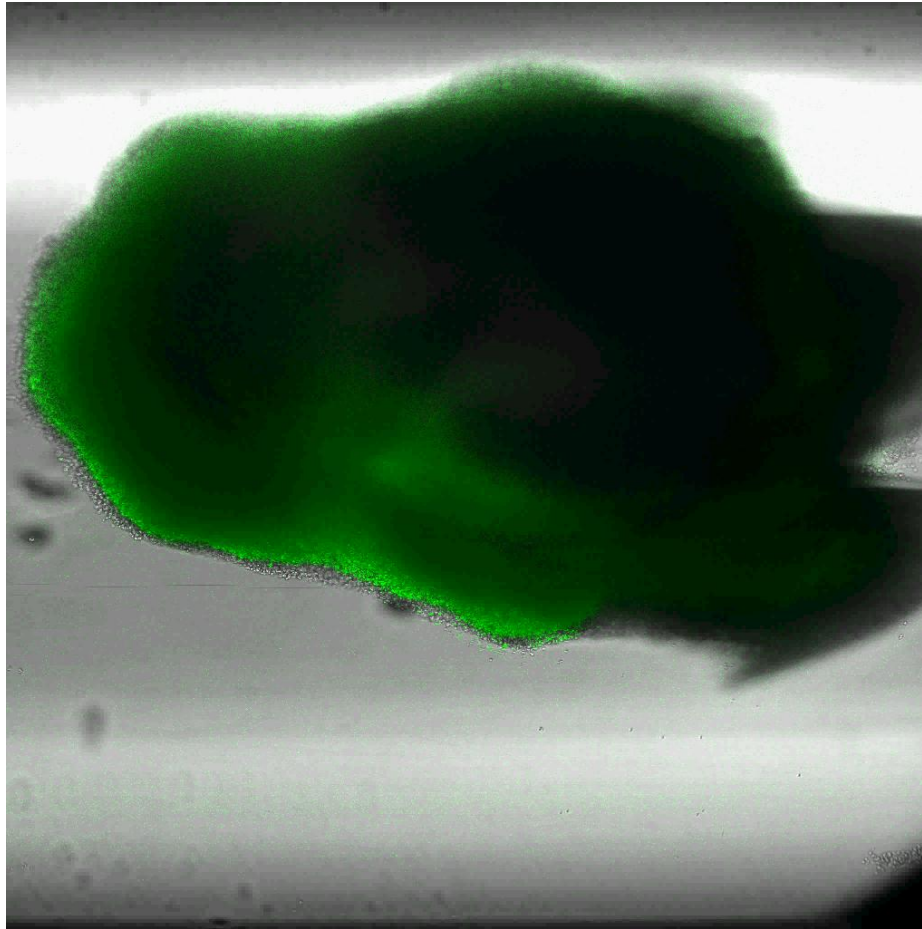
60 min

Treatment with 50 mg/l Chlorine



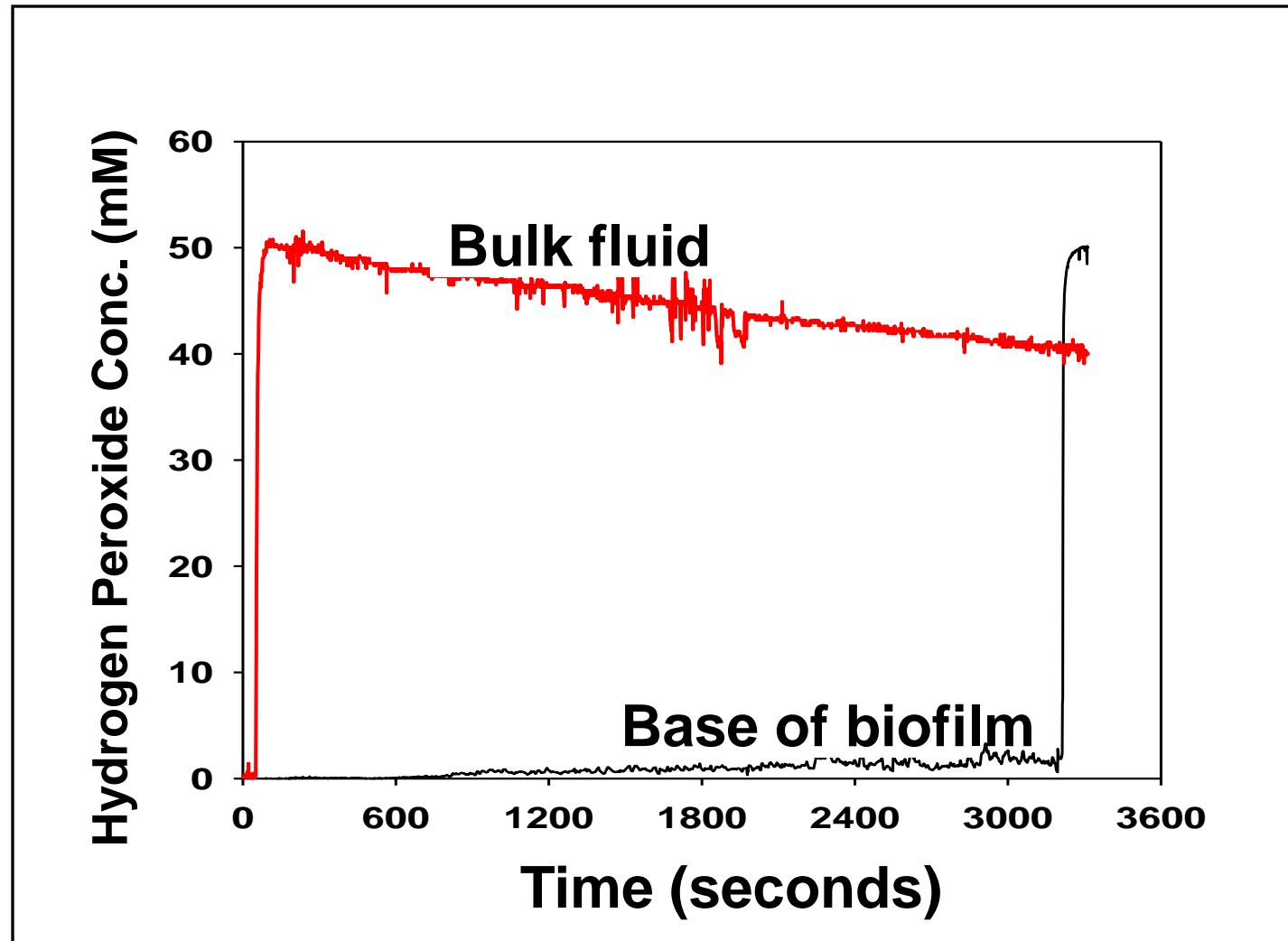
60 min

Treatment with 10 mg/l Chlorine Dioxide



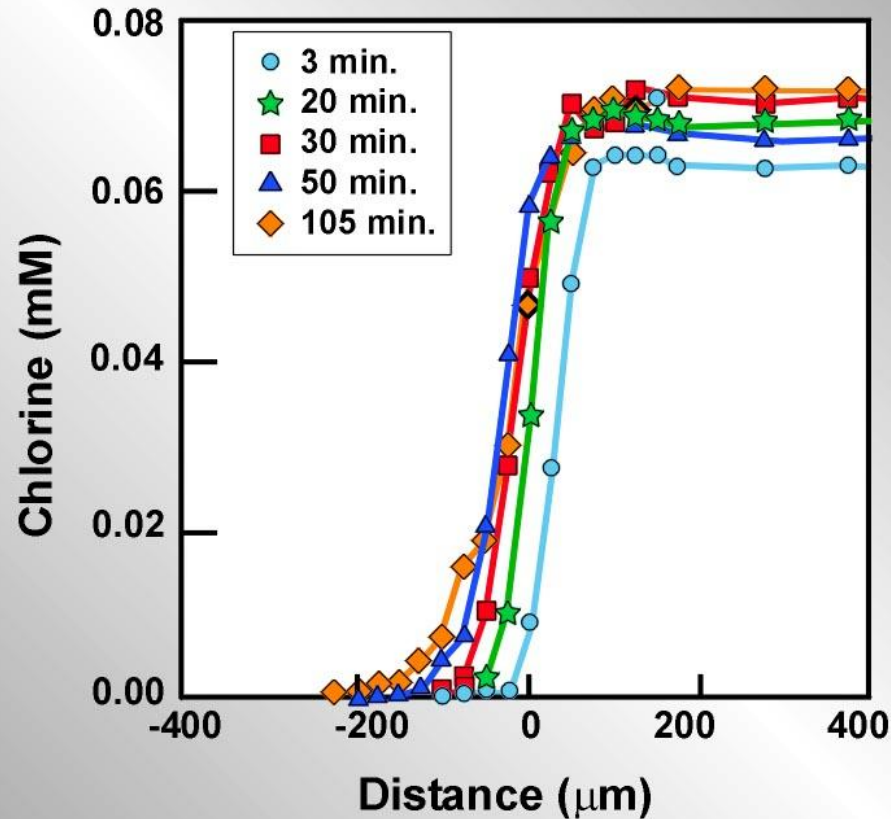
60 min

Hydrogen Peroxide cannot penetrate Biofilm



Stewart et al (2000) *Appl. Environ. Microbiol.* **66**:836.

Chlorine cannot penetrate Biofilm



2642-89cs

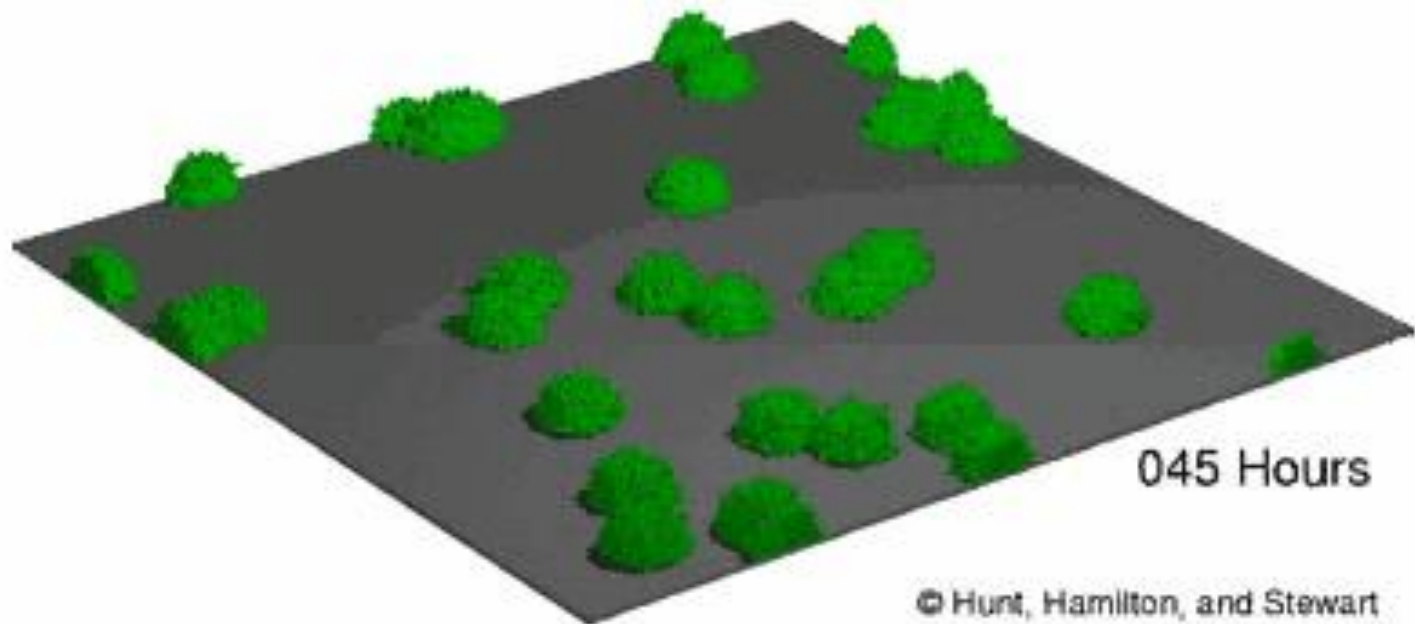
de Beer et al (1994) *Appl. Environ. Microbiol.* **60**:4339.

Oxidizer Potential

- Oxidation Strength describes the strength of the reaction and reactivity. A higher number means it reacts with more things.
- Oxidation Capacity indicates how many electrons are transferred.

Oxidizer	Oxidation Strength	Oxidation Capacity
Ozone (O ₃)	2.07	2 e ⁻
Hydrogen Peroxide (H ₂ O ₂)	1.78	2 e ⁻
Hypochlorous Acid (HOCl)	1.49	2 e ⁻
Hypobromous Acid (HOBr)	1.33	2 e ⁻
Chlorine Dioxide (ClO ₂)	0.95	5 e ⁻

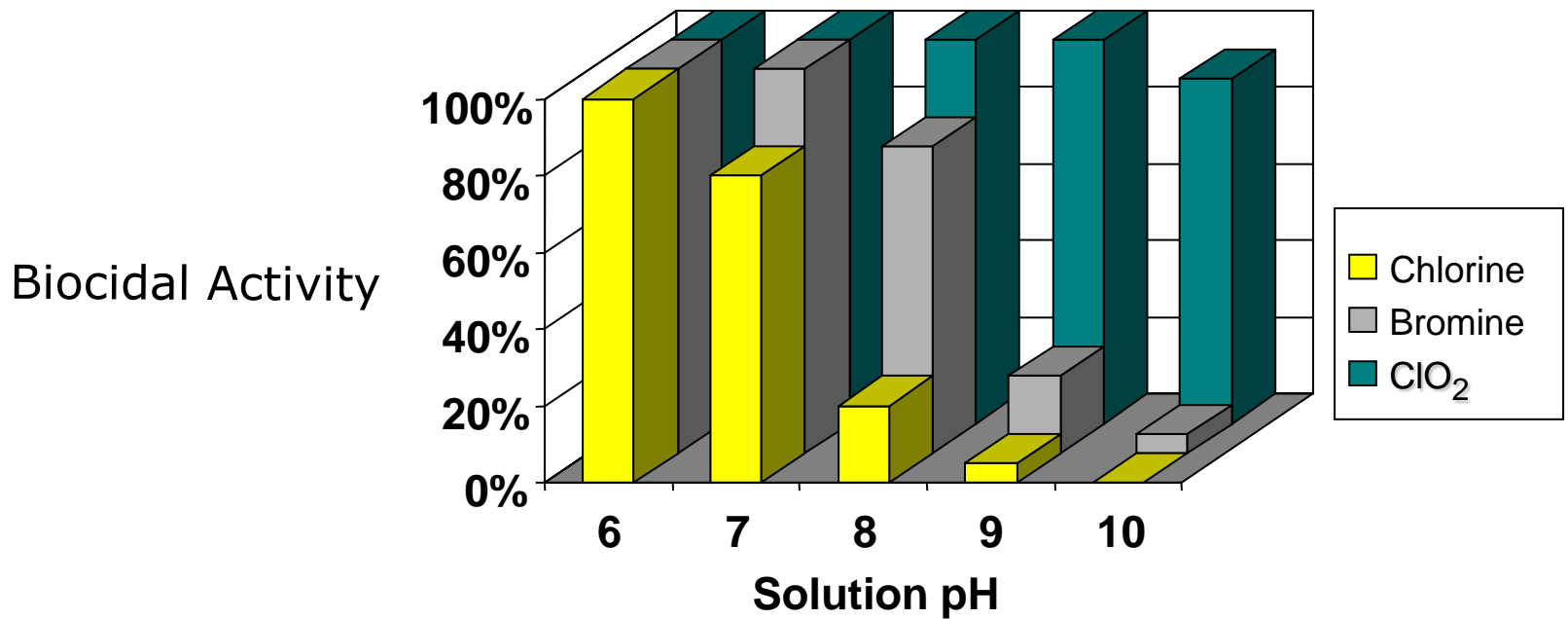
Regrowth of Biofilms



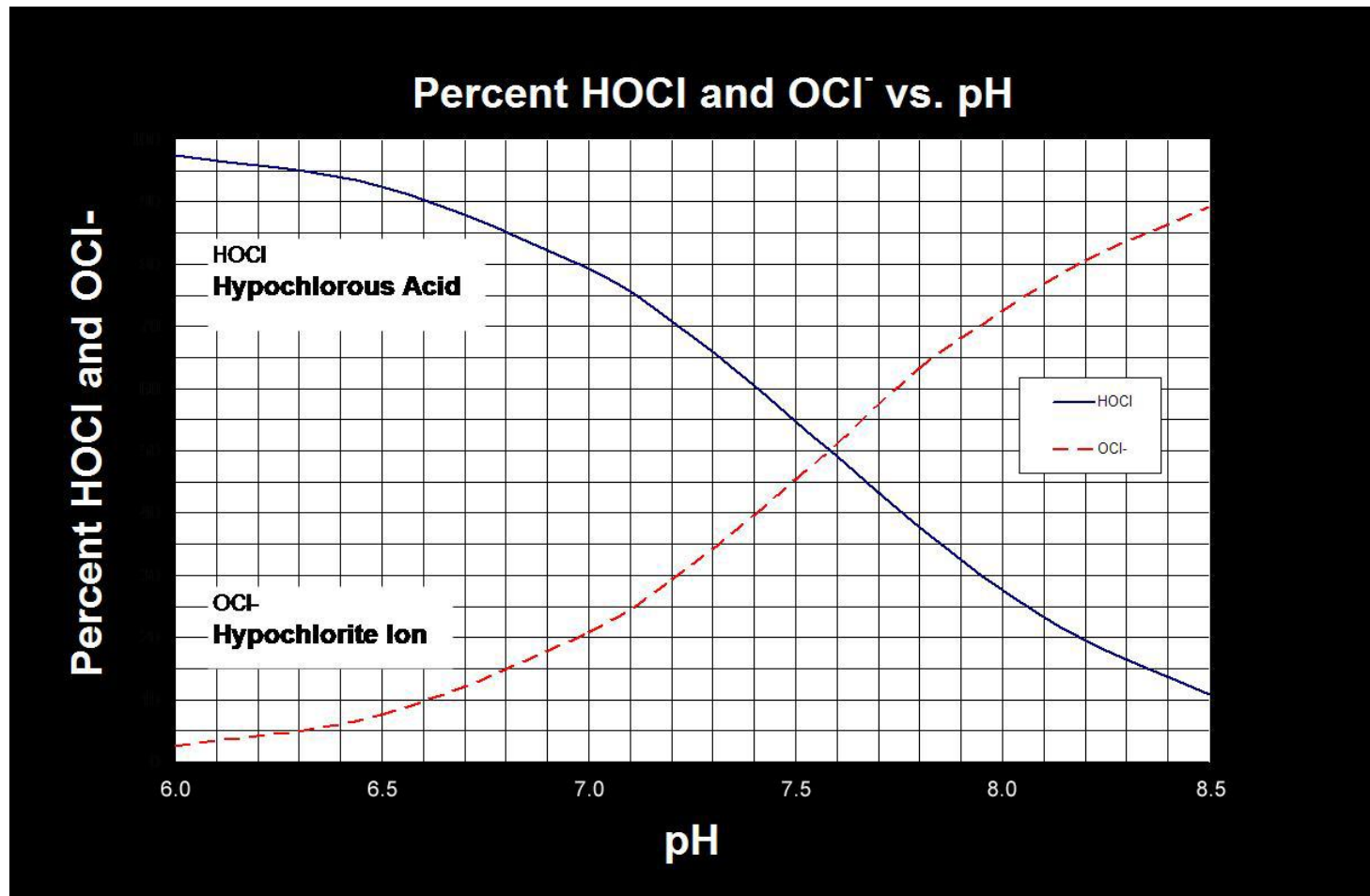
Regulation of Biofilms

- Biofilms are considered a “pest” by the EPA
- Biofilm is regulated under FIFRA
 - ▣ Federal Insecticide Fungicide Rodenticide Act
- Currently there are no validated methods for the disinfection efficacy of biofilms
- EPA has started the process to validate a biofilm efficacy test method

Chlorine Dioxide is not pH Dependent

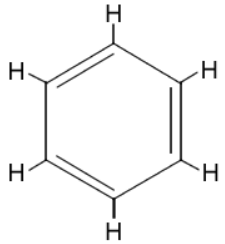


pH Effects on Active Chlorine

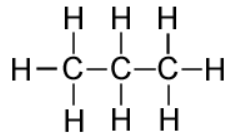


Chlorine Harmful Byproducts

- Tri Halo Methane (THMs)
 - Chlorine, Bromine, Iodine

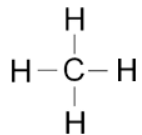


Aromatics

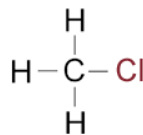
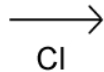


Chain

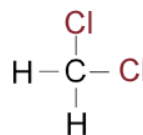
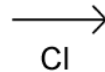
Chlorine Substitution Reaction
Harmful by products
Trihalomethane - THMs



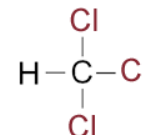
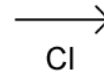
Methane



Mono-



Di-



Trichloromethane

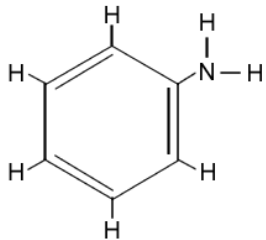


Chloroform (gas)

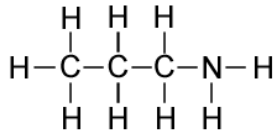
Chlorine Harmful Byproducts

□ Chloramines – Chlorine, Bromine, Iodine

▣ Ammonia, Nitrates, Proteins

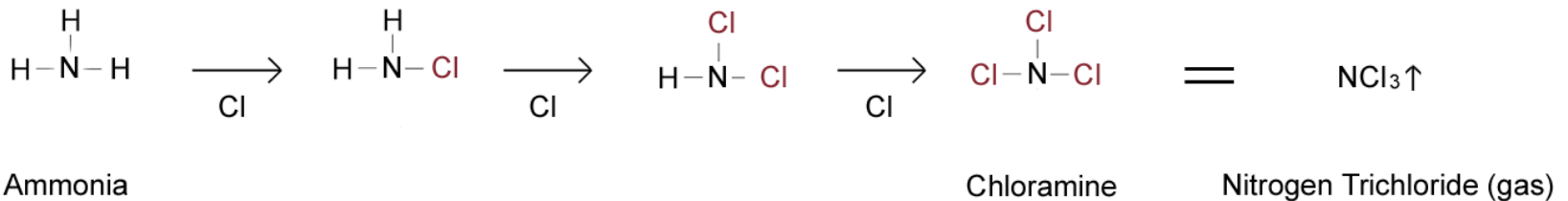


Aromatics

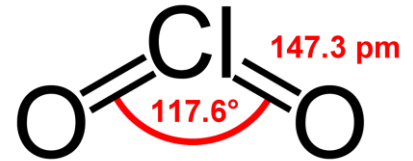
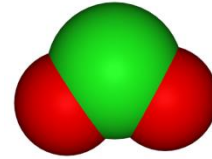


Chain

Chlorine Substitution Reaction
Harmful by products
Chloramines



Chlorine Dioxide



- Powerful Water Sanitizer
- Oxidizer 2.5x more effective than Chlorine
- 10x more soluble in water than chlorine
- Low Sodium, Low Chlorite
- No hydrophilic substitution reactions
- Does not form THMs
- Does not form Chloramines
- Effective at wide pH ranges

Biofilm Removal by ClO_2

- Chlorine Dioxide penetrates biofilm

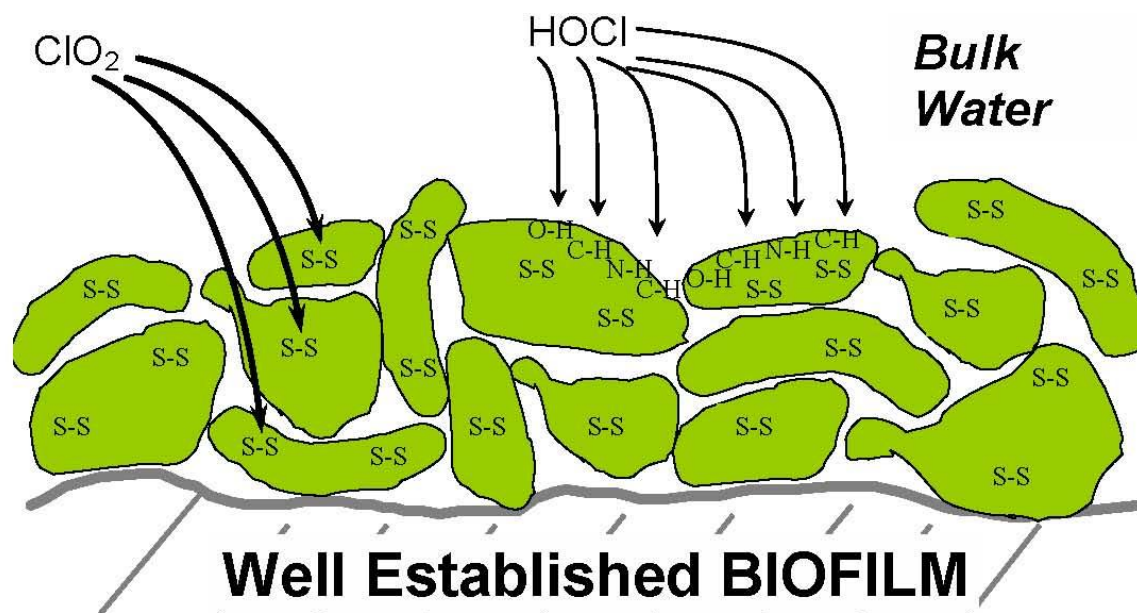
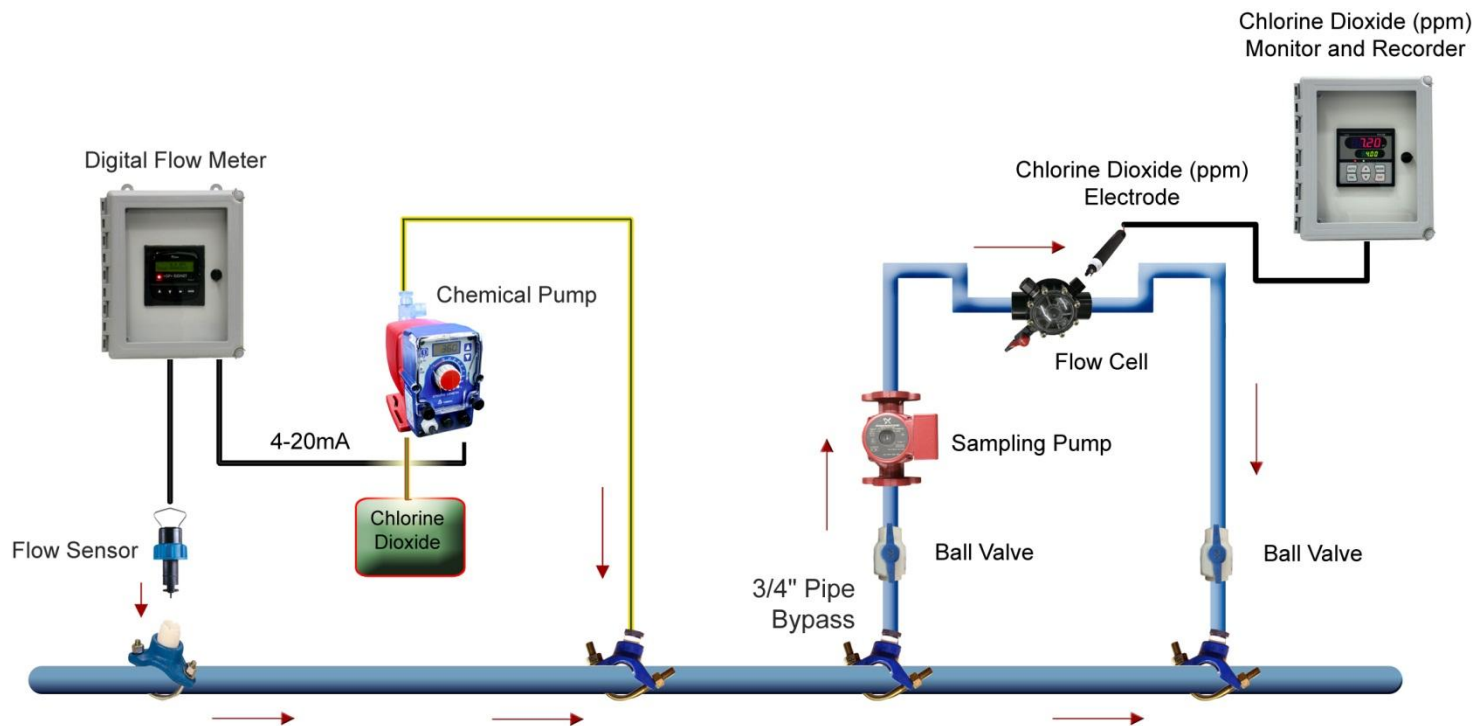


Figure 1. Effect of chlorine and chlorine dioxide at biofilm penetration.

<http://postharvest.tfrec.wsu.edu/PC2004B.pdf>

Douglas G. Kelley, Ph. D. Technical Director

Simple Chemical Injection



Chlorine Dioxide Generators



2 Chemical Process – 80% efficient



3 Chemical Process – 100% efficient



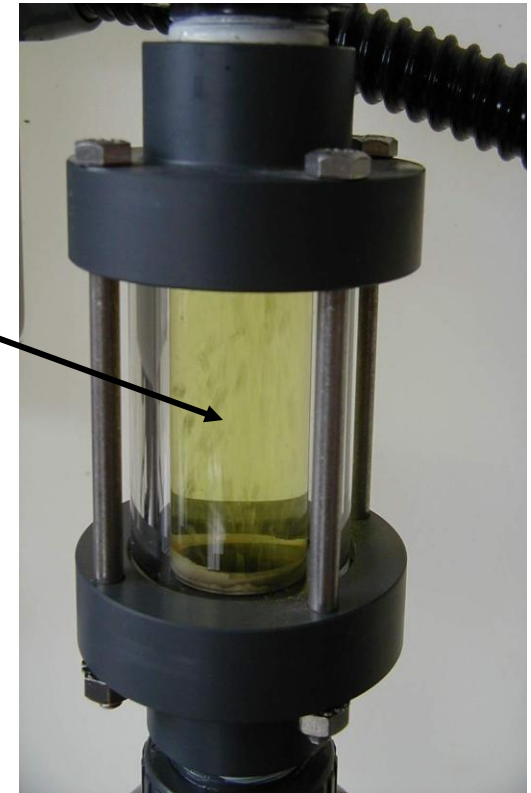
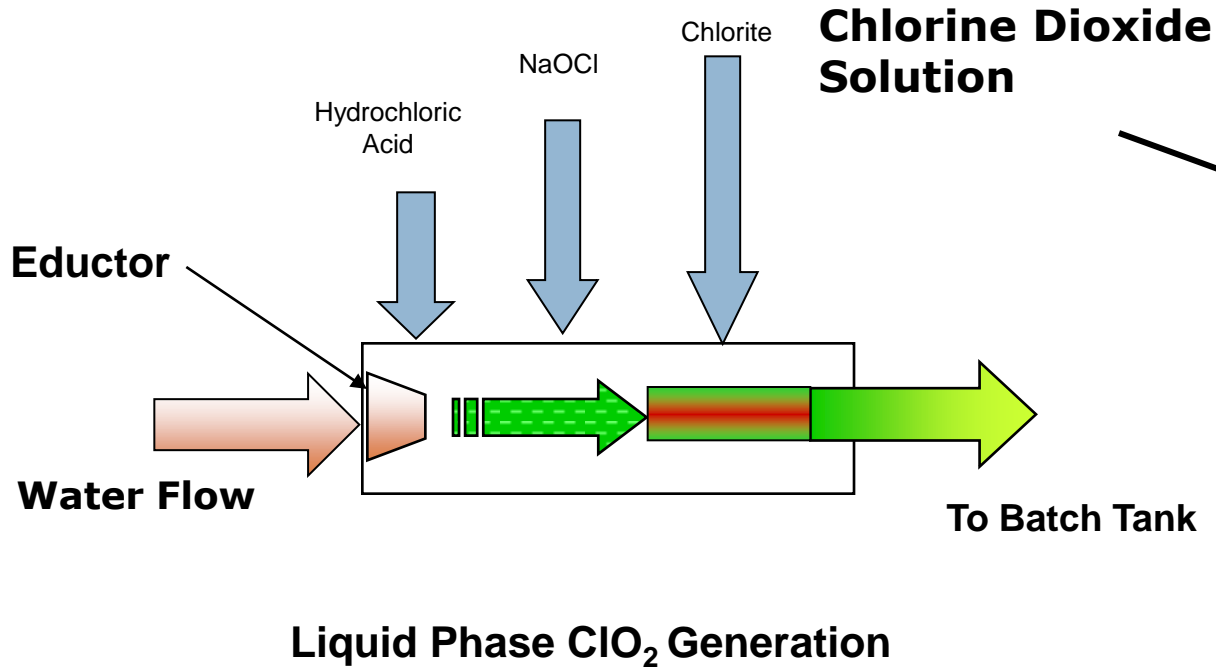
Chlorine Dioxide Oxidation



Chlorine Dioxide System



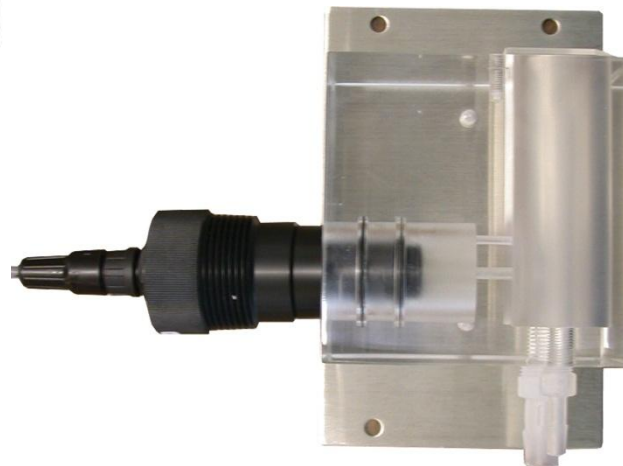
Production of Chlorine Dioxide



Monitoring Treatment



- Chlorine Dioxide Monitoring (ppm)
- Membrane Sensor
- Amperometric Measurement



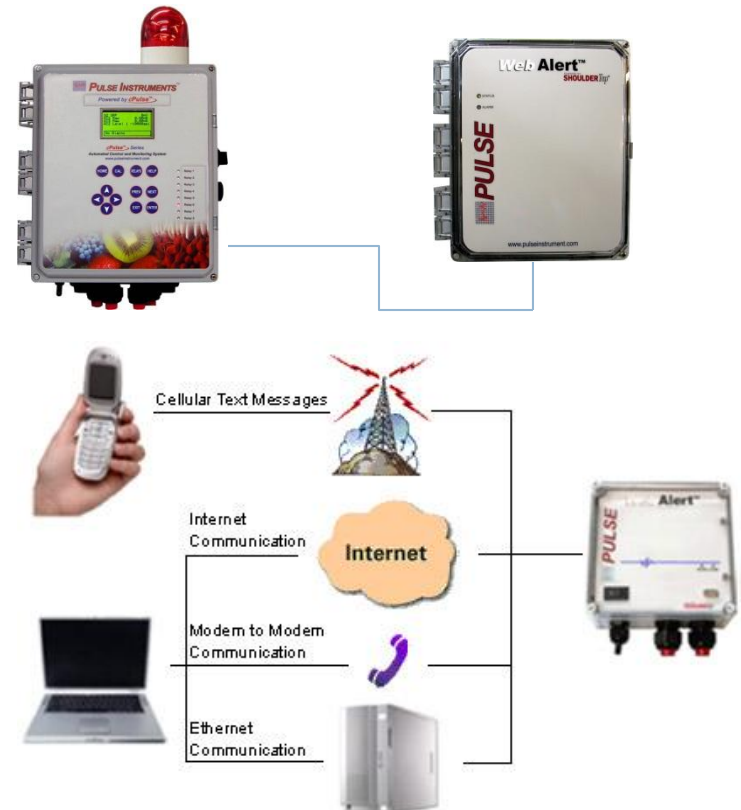
APS Treatment Program



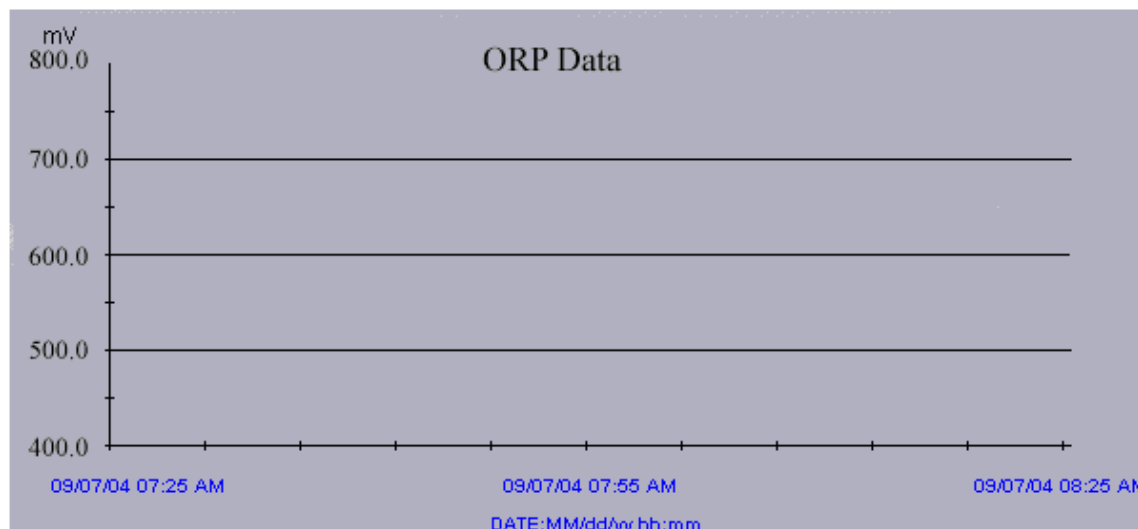
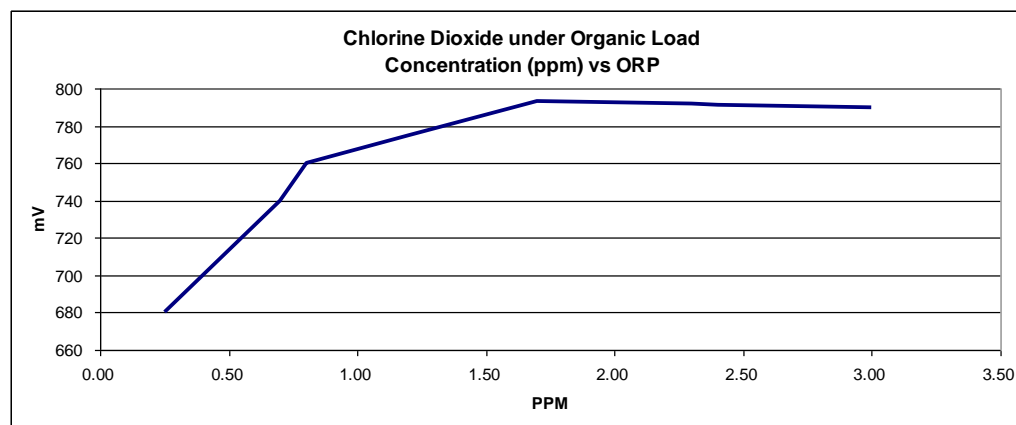
APS Full Service Program

- ❑ Full Service Program
- ❑ APS Generator
- ❑ Maintenance and Service
- ❑ Includes Chemicals
- ❑ Includes Freight
- ❑ Process Monitoring
- ❑ Real Time Access
- ❑ Data Recording and Reports
- ❑ Data Management

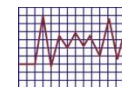
Data Recording and Management



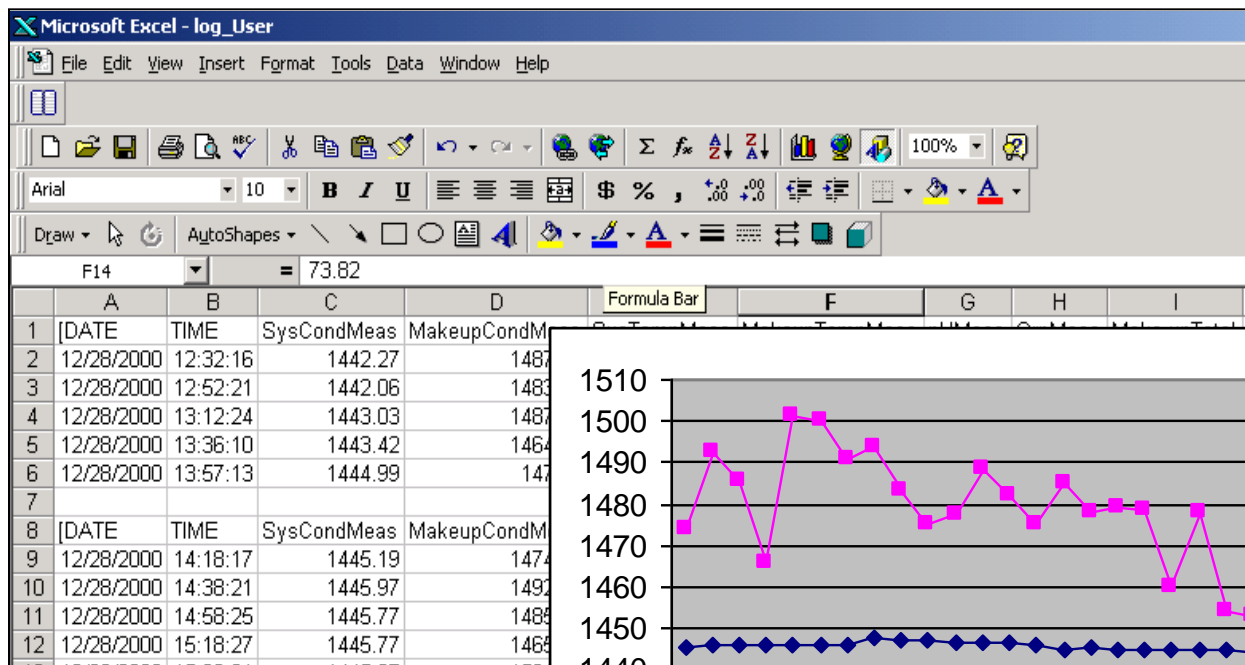
Chlorine Dioxide ORP



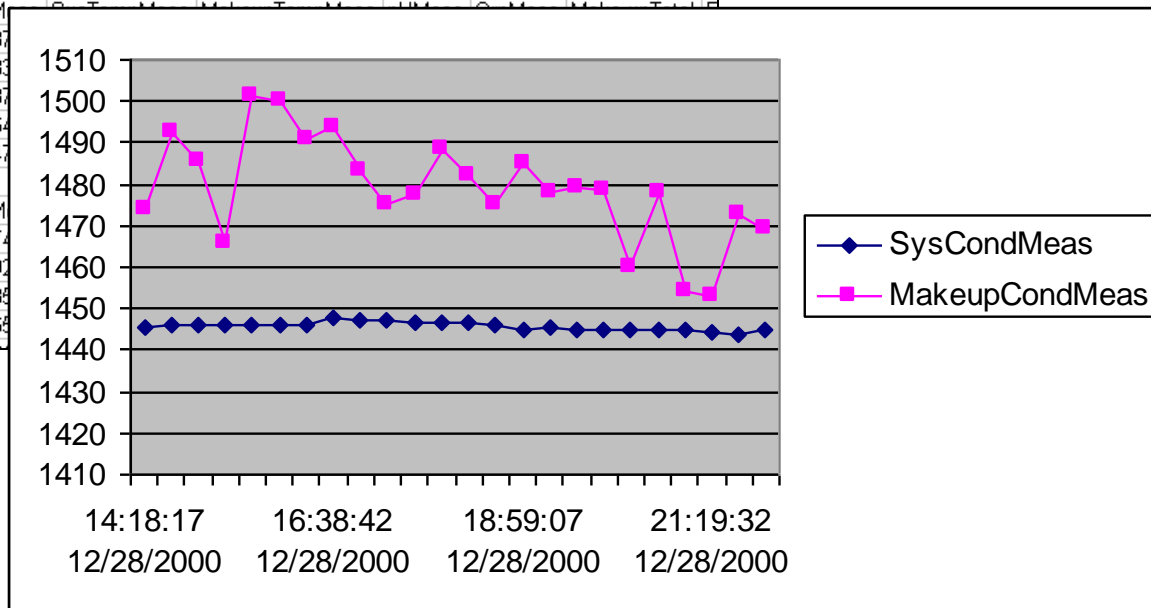
View Download Data File in Excel



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Ongoing Research Affiliation



- University of Florida – Dr. Paul Fisher
 - ▣ Water Education Alliance Member
 - ▣ Young Plant Research Center
- Konjoian Research and Education
 - Dr. Peter Konjoian
- UC Davis – At “The Bubble” and Greenhouse Production
 - Dr. Michael Parrella
- UC Davis – Fresh Produce
 - Dr. Trevor Suslow
- Washington State University – Tree Fruit Research
 - Dr. Karen Killinger
- Montana State University – Biofilm Engineering
 - Dr. Phil Stewart

